Heterocyclic Transformations by Hydrogenolysis. Pyrimidine Derivatives from 1,2,4-Oxadiazole Derivatives

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As described in previous communications (1) it is possible to obtain a series of heterocyclic compounds by rearrangement of a suitable substituted 1.2,4-oxadiazole system. Starting from Palazzo and coworkers (2) observation that reductive catalytic hydrogenation of the 1,2,4-oxadiazole ring containing alkyl and aryl groups at position 3 and 5 yielded acylamidines, our attention has now been directed to the use of the 1,2,4-oxadiazole system, containing at position 3 a side-chain able to give ring closure, as a precursor of different heterocyclic systems through the reductive cleavage of N-O bond. In this communication we report a novel synthesis of the pyrimidine ring system as the first example of transformation of 1,2,4-oxadiazole system by hydrogenation.

Hydrogenation of N-(1,2,4-oxadiazol-3-yl) β -enamino esters (la-d) (3) on Pd/C in ethanol yielded 2-acylamino-pyrimidones derivatives (IIIa-d) which must be considered as arising directly from intermediates (IIa-d) resulting from 1,2,4-oxadiazole ring opening.

Under the same experimental condition N-(1,2,4-oxadiazol-3-yl) β -enaminoketones (IVa-b) (3,4) yielded directly 2-acetamidopyrimidines derivatives (Vla-b), whereas enaminoketones (IVc-d) (3,4) afforded an oily residue (several spots on tlc), containing probably the intermediates

(Vc-d), which, without purification, by treatment with aqueous sodium hydroxide gave, beside benzoic acid, 2-aminopyrimidines (VIIc-d).

The structures of compounds IIIa-d, VIa-b and VIIc-d were determined by elemental analyses, spectroscopic evidence (nmr) and, in known cases, by direct comparation with authentic samples.

EXPERIMENTAL

All melting points (Kofler) are uncorrected; nmr (DMSO-d₆): Jeol C-60H spectrometer (TMS as internal reference).

Hydrogenation of the N(1,2,4-Oxadiazol-3-yl)- β -enamino Esters (Ia-d) and N(1,2,4-oxadiazol-3-yl)- β -enaminoketones (IVa-d). General Procedure.

A mixture of 0.01 mole of the compounds (la-d) or (IVa-d), 200 ml. of ethanol and 1 g. of 5% Pd/C was hydrogenated in a Parr apparatus at 40 psi for 4 hours at room temperature. Removal of

the catalyst and evaporation of ethanol left the reduced products, yield, $ca.\,50\text{-}60\%$.

(a) Hydrogenation of la: 2-Acetamido-6-methylpyrimidin-4-one (IIIa).

The product melted at 222°, (ethanol); lit. (5) m.p. 220-221°; nmr: 2.17 δ (s, 6H, 2 x CH₃), 5.98 δ (s, 1H, C₅-H), 11.68 δ (br. s, 2H, NH) (6).

(b) Hydrogenation of Ib: 2-Acetamido-6-phenylpyrimidin-4-one (IIIb).

The product melted at 254°, (ethanol); lit. (7) m.p. 254-255°; nmr: 2.20 δ (s, 3H, CH₃), 6.66 δ (s, 1H, C₅-H), 7.40-8.20 δ (m, 5H, Ar-H), 11.76 δ (br. s, 2H, NH).

(c) Hydrogenation of Ic: 2-Benzamido-6-methylpyrimidin-4-one (HIc).

The product melted at 198°, (ethanol); nmr: 2.24 δ (s, 3H, CH₃), 5.98 δ (s, 1H, C₅-H) (6), 7.40-8.40 δ (m, 5H, Ar-H), 12.20 δ (br. s, 2H, NH).

Anal. Calcd. for $C_{12}H_{11}N_3O_2\colon C,62.87;\ H,4.84;\ N,18.33.$ Found: $C,62.73;\ H,4.79;\ N,18.44.$

(d) Hydrogenation of ld: 2-Benzamido-6-phenylpyrimidin-4-one (IIId).

The product melted at 230°, (ethanol); nmr: 6.65 δ (s, 1H, C₅-II), 7.35-8.20 δ (m, 10H, Ar-H), 11.90 δ (br. s, 2H, NH). Anal. Calcd. for C₁₇H₁₃N₃O₂: C, 70.09; H, 4.50; N, 14.43. Found: C, 70.24; H, 4.54; N, 14.35.

(e) Hydrogenation of IVa: 2-Acetamido-4,6-dimethylpyrimidine (Vla).

The product melted at 123°, (water); lit. (8) m.p. 124°; nmr: 2.15 δ (s, 3H, CH $_3$ -CO-NH-), 2.30 δ (s, 6H, 2 x CH $_3$), 6.85 δ (s, 1H, C $_5$ -H), 10.17 δ (s, 1H, NH).

(f) Hydrogenation of IVb: 2-Acetamido-4-methyl-6-phenylpyrimidine (VIb).

The product melted at 148°, (ethanol); lit. (9) m.p. 146°; nmr: 2.32 δ (s, 3H, CH₃-CO-NH-), 2.50 δ (s, 3H, CH₃), 7.58-7.82 δ (m, 4H, meta and para Ar-H), 8.15-8.50 δ (m, 2H, ortho Ar-H), 10.44 δ (s, 1H, NH).

(g) Hydrogenation of IVc: 2-Amino-4,6-dimethylpyrimidine (VIIc).

Removal of the catalyst and evaporation of ethanol left an oil which was refluxed (30 minutes) with 20% aqueous sodium hydroxide (2 ml.) and ethanol (20 ml.). After removing the solvent, water (10 ml.) was added and, after cooling, VIIc was obtained, yield, 50%, m.p. 154° , (water); lit. (10) 152° ; nmr: $2.12~\delta$ (s, 6H, 2~x CH₃), $6.28~\delta$ (s, 1H, C_5 -H), $6.37~\delta$ (s, 2H, NH₂).

Acidification of the alkaline solution gave benzoic acid.

(h) Hydrogenation of IVd: 2-Amino-4-methyl-6-phenylpyridine (VIId).

As described for the previous sample; m.p. 176° , (ethanol); lit. (9) m.p. 173° ; nmr: 2.34 δ (s, 3H, CH₃), 6.66 δ (s, 2H, NH₂), 7.05 δ (s, 1H, C₅-H), 7.45-8.30 δ (m, 5H, Ar-H).

Acidification of the alkaline solution gave benzoic acid.

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